

C2 Exothermic and Endothermic Quiz

1. How can you show that a reaction is exothermic?
2. What is the meaning of endothermic?
3. In an endothermic reaction the temperature of the surroundings drops. Why?
4. Give examples of endo and exothermic reactions.
5. If a forward reaction is exothermic, what do you need to do to reverse the reaction?
6. How do you know from an energy profile diagram that a reaction is exothermic?
7. How do you know from an energy profile diagram that a reaction is endothermic?
8. How do you convert blue hydrated copper sulphate into white anhydrous copper sulphate?
9. Give a use for anhydrous copper sulphate.